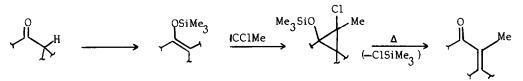
REACTION OF CHLOROMETHYLCARBENE WITH TRIMETHYLSILYL ENOL ETHERS. SYNTHESIS OF EUCARVONE, (±)-NUCIFERAL AND (±)-MANICONE. By L. Blanco, N. Slougui, G. Rousseau and J.M. Conia Laboratoire des Carbocycles*, Université de Paris-Sud, Bâtiment 420 91405 ORSAY, France

Summary. New synthesis of Eucarvone $\underline{7}$, $(\underline{*})$ -Nuciferal $\underline{12}$ and $(\underline{*})$ -Manicone $\underline{16}$ are described from trimethylsilyl enol ethers $\underline{2}$, $\underline{10}$ and $\underline{14}$ via their chloromethylenation products by means of a two carbons homologation reaction.

As a part of our synthetic program from silyl enol ethers and carbenes and carbenoids, i.e methylenation by a modified Simmons-Smith reaction (1) and dihalomethylenation (2), we recently reported the addition of chloromethylcarbene to silyl enol ethers which, followed by thermal rearrangement, constitutes a convenient way to α -methyl α -ethylenic compounds (3).

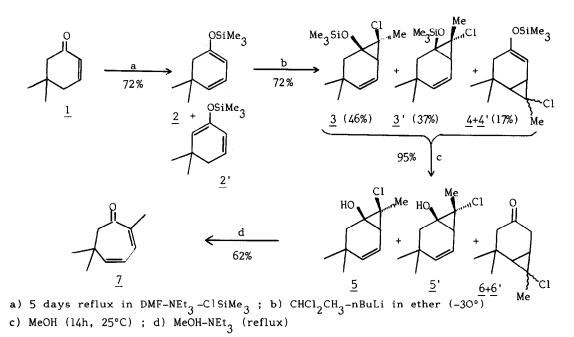


The purpose of the present communication is to show that such a pathway can be conveniently applied to the synthesis of natural products containing the α -methyl α -ethylenic carbonyl pattern ; for instance Eucarvone <u>7</u>, ($\frac{1}{2}$)-nuciferal <u>12</u> and ($\frac{1}{2}$)-manicone <u>16</u>.

Synthesis of Eucarvone 7

To our knowledge, there is only one report concerning the total synthesis of Eucarvone 7 carried out by Barnes and Houlihan (4), from 2,2-dimethylcyclohexanone, in low yield. As reported in scheme 1, our synthesis started from the readily available 5,5-dimethylcyclohexene-2-one 1,which was converted, following the standard procedure of House (5), into a 85:15 mixture of trimethylsilyl enol ether 2 and its kinetic isomer 2' (yield : 72%). Diene 2, purified by distillation (yield : 61% from 1 ; b.p. 71°/12 mmHg) (6) was treated with chloromethylcarbene generated in situ by a slow addition (7 hours) of n-butyllithium to 1,1-dichloroethane in ether. A mixture of four products was obtained : the expected chloromethyltrimethylsiloxycyclopropanes 3, 3' (83%), beside the two trimethylsilyl enol ethers 4, 4'. All these

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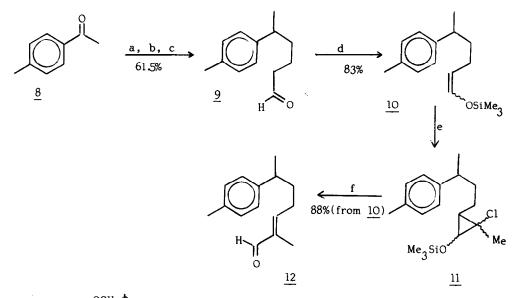


Scheme 1 - Synthesis of Eucarvone 7

products were identified by glc-ms analysis. Upon methanolysis the mixture 3, 3', 4, 4' gave two alcohols 5, 5' (78.5%), readily separated (ratio 55:45) by liquid chromatography (ether-pentane) on silica gel (6), and two ketones 6, 6' (16%). The structure of alcohols 5 and 5' was determined by their IR spectra : 5 (Cl and OH cis) showed a strong intramolecular H bond, which was absent for the isomer 5' (Cl and OH trans). The later, on reflux in the mixture MeOH-NEt₃ (85:15), led after one day to Eucarvone 7 (yield : 86%), while, on reflux in the same conditions for one week, the epimeric alcohol 5 underwent the conversion into 7 in 40% yield only. This difference of behaviour between 5 and 5' gives rise to interesting mechanistic problems, now under investigation. Eucarvone 7 was finally obtained in an overall yield of 30% from ketone 1.

Synthesis of (\pm) -Nuciferal 12

Sesquiterpene Nuciferal was exceeding studied (7). Our synthesis (scheme 2) was carried out through aldehyde <u>9</u>, easily obtained in a three steps process from p.methylacetophenone <u>8</u> ((10), yield : 61.5%). Then, the standard procedure of House (5) led to the trimethylsilyl enol ethers <u>10</u> (10) (Z : E mixture (42:58)) which underwent quantitatively chloromethylcarbene addition to give the chloromethyl-trimethylsiloxycyclopropane isomers <u>11</u> (identified from the mixture by glc-ms analysis and by solvent effect in NMR). On heating in methanol-triethylamine <u>11</u> gave pure (±)-Nuciferal 12 (yield : 88% from 10).

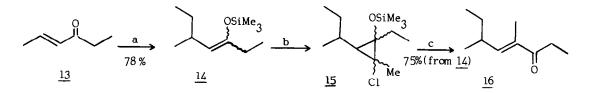


a) 1°) Cl $\sim OCH_2 \Phi$ (8) /Mg/Et₂O (20°C) ; 2°) H₃O⁺. b) H₂-Pd/C (MeOH, HCl conc) c) (COCl)₂ -DMSO in CH₂Cl₂(9). d) l day reflux in DMF-NEt₃ -ClSiMe₃ (under N₂) e) CHCl₂-CH₃/n-BuLi in ether (-30°C). f) MeOH-NEt₃ (l.5 days reflux).

Scheme 2 : Synthesis of (±)-Nuciferal 12

Synthesis of (±)-Manicone 16

Manicone, an alarm pheromone of certain species of <u>Manica</u> ants, has been identified as (E) 4,6-dimethyl-4-octen-3-one <u>16</u> (11). Our synthesis (scheme 3) started from the readily available 4-h, ken-3-one <u>13</u> (12), which by a 1,4-c ddition of EtMgBr, followed by quenching of the enolate with ClSiMe₃, led to the trimethylsilyl enol ethers <u>14</u> (Z:E mixture (30:70)) (13). Then, the addition of chloromethylcarbene to <u>14</u> led to the isomeric chloromethyltrimethylsiloxycyclopropanes <u>15</u> (analysis carried out by glc.ms). On heating in toluene, the mixture of <u>15</u> gave pure (\pm)-Manicone 16 (yield : 75% from <u>14</u>).

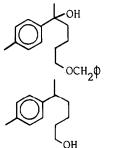


a) 1°) EtMgBr, 0.5 CuI (Et₂O, -60° C) ; 2°) ClSiMe₃ (20°C) (in the presence of HMPA) b) CHCl₂CH₃/n-BuLi in ether (-30° C). c) Toluene (10 days reflux).

Scheme 3 : Synthesis of (±)-Manicone 16

REFERENCES

- J.M. Conia and C. Girard, <u>Tetrahedron Letters</u>, 2767 (1973); C. Girard and J.M. Conia, <u>Tetrahedron Letters</u>, 3327 (1974); 3333 (1974); C. Girard, P. Amice, J.P. Barnier and J.M. Conia, <u>Tetrahedron Letters</u>, 3329 (1974).
- 2) P. Amice, L. Blanco and J.M. Conia, Synthesis, 196 (1976).
- 3) L. Blanco, P. Amice and J.M. Conia, Synthesis, in the press.
- 4) R.A. Barnes and W.J. Houlihan, <u>J. Org. Chem.</u>, <u>26</u>, 1609 (1961).
- 5) H.O. House, L.J. Czuba, M. Gall and H.D. Olmstead, <u>J. Org. Chem.</u>, <u>34</u>, 2324 (1969).
- 6) 2 ⁴HNMR (CCl₄) § (ppm) : 0.20 (s, 9H), 1.11 (s, 6H), 2.05 (d, 2H), 5.02 (m, 2H), 5.62 (dd, 1H) ; IR (CCl₄) (cm⁻¹) : 3040, 1650, 1590. 5 ⁴HNMR (CCl₄) § (ppm) : 1.00 (s, 3H), 1.15 (s, 3H), 1.46 (s, 3H), 1.00 to 2.00 (m, 3H), 5.70 (m, 2H) ; IR (CCl₄) (cm⁻¹) : 3580 (Intramol. H bond), 3030, 1670. 5' ⁴HNMR (CCl₄) § (ppm) : 1.02 (s, 3H), 1.12 (s, 3H), 1.75 (s, 3H), 1.00 to 2.00 (m, 3H), 5.65 (m, 2H) ; IR (CCl₄) (cm⁻¹) : 3600 (free H bond), 3420 (Intermol. H bond), 3030, 1660. 6 + 6' ⁴HNMR (CCl₄) § (ppm) : 1.10 and 1.15 (s, 3H), 1.25 (s, 3H), 1.68 and 1.76 (s, 3H), 1.00 to 2.80 (m, 6H) ; IR (CCl₄) (cm⁻¹) : 1720 (4 C=0). 7 ⁴HNMR (CCl₄) § (ppm) : 1.16 (s, 6H), 1.88 (s, 3H), 2.62 (s, 2H), 5.55 to 6.55 (m, 3H) ; IR (CCl₄) (cm⁻¹) : 1655 (4 C=0).
- 7) T. Nakai, H. Shiono and M. Okawara, <u>Chem. Letters</u>, 249 (1975) ; K. Yamamoto, J. Yoshitake, N.T. Qui and J. Tsuji, <u>Chem. Letters</u>, 859 (1978) ; Y. Masaki, K. Hashimoto, K. Sakuma and K. Kaji, <u>J.C.S.Chem. Comm.</u>, 855 (1979) ; K. Kondo and D. Tunemoto, <u>Tetrahedron Letters</u>, 1007 (1975) and references therein.
- 8) G.M. Bennett and A.L. Hock, J. Chem. Soc., 472 (1927).
- 9) A.J. Mancuso, S.L. Huang and D. Swern, J. Org. Chem., 43, 2480 (1978).



⁴ HNMR (CCl₄) & (ppm) : 1.40 (s, 3H), 2.25 (s, 3H), 0.80 to 2.50 (m, 7H), 3.30 (t, J = 6 Hz, 2H), 4.22 (d, 2H), 7.10 (m, 4H) ; IR (CCl₄) (cm⁻¹) : 3597 (free H bond).

⁴HNMR (CC1₄) & (ppm) : 1.18 (d, J = 7 Hz, 3H), 0.95 to 1.85 (m, 6H), 2.26 (s, 3H), 2.35 to 2.80 (m, 1H), 3.00 (bs, 1H (OH)), 3.40 (bt, J = 6 Hz, 2H), 6.98 (s, 4H); IR (CC1₄) (cm⁻¹) : 3635 (free H bond).

9 ⁴HNMR (CC1₄) δ (ppm) : 1.20 (d, J = 7 Hz, 3H), 1.35 to 1.70 (m, 4H), 2.26 (s, 3H), 2.05 to 2.90 (m, 3H), 6.98 (s, 4H), 9.58 (t, J = 1.5 Hz, 1H) ; IR (CC1₄) (cm⁻¹) : 2705, 1900, 1725 (\checkmark C=0), 1510, 1450. 10 ⁴HNMR (CC1₄) δ (ppm): (Z+E) 0.09 (s, 9H), 1.20 (d, J = 7 Hz, 3H), 0.90 to $\overline{2.30}$ (m, 4H), 2.25 (s, 3H), 4.20 to 6.20 (m, 2H), 6.98 (s, 4H).

- 11) O.P. Vig, S.D. Sharma, S.D. Kumar and V.K. Handa, Indian J. Chem., 16B, 740 (1978); T. Mimura, Y. Kimura and T. Nakai, <u>Chem. Letters</u>, 1361 (1979) and references therein.
- 12) Obtained by isomerisation of 5-hexen-3-one ; see G. Rousseau and J.M. Conia, following communication.
- 13) 14 ⁴HNMR (CC14) δ (ppm) : 0.15 (s, 9H), 0.60 to 2.70 (m, 14H), 4.08 (d, J = 10 Hz, 1H) (isomer E), 4.12 (d, J = 10 Hz, 1H) (isomer Z) ; IR (CC14)(cm⁻¹) 2960, 1662 (∂ C=C), 840.

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